

Chemistry 3502/4502**Exam II****February 27, 2006**

- 1) Circle the correct answer on multiple-choice problems.**
- 2) There is *one* correct answer to every multiple-choice problem. There is no partial credit. On the short-answer problem, show your work in full.**
- 3) A table of useful integrals and other formulae is provided at the end of the exam.**
- 4) You should try to go through all the problems once quickly, saving harder ones for later.**
- 5) There are 10 multiple-choice problems. Each is worth 8 points. The short-answer problem is worth 20 points.**
- 6) There is no penalty for guessing.**
- 7) Please write your name at the bottom of each page.**
- 8) Please mark your exam with a pen, not a pencil. If you want to change an answer, cross your old answer out and circle the correct answer. Exams marked with pencil or correction fluid will not be eligible for regrade under any circumstances.**

Score on Next Page after Grading

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1. For the diatomic molecule CD, where C has atomic mass 12 and D has atomic mass 2, what is the reduced mass?

- | | |
|------------|---|
| (a) 7 / 12 | (e) 10 |
| (b) 12 / 7 | (f) It depends on the vibrational state |
| (c) 1 / 7 | (g) Cannot be determined from information given |
| (d) 2 / 7 | (h) None of the above |

2. Which of the following statements is/are *true* for a given set of QMHO wave functions corresponding to the same harmonic potential V ?

- | | |
|--|---|
| (a) The ground state energy is zero, i.e., the bottom of the potential | (e) The wave functions are eigenfunctions of the parity operator |
| (b) The number of nodes is equal to $n+1$, where n is the energy level | (f) The selection rule for spectroscopic transitions is $n \rightarrow n \pm 2$ |
| (c) $\langle T \rangle_n = \langle V \rangle_n = \langle E \rangle_n$ | (g) (c), (e), and (f) |
| (d) The wave functions have zero amplitude beyond the classical turning points | (h) All of the above |

3. Which of the following statements about angular momentum operators and their eigenvalues and eigenfunctions is/are *true*?

- | | |
|--|---|
| (a) $L_+ = -(L_-)^*$ | (e) $[L_x, L_y] = 2\hbar L_z$ |
| (b) $\langle L^2 \rangle = \langle L_z \rangle^2$ whenever $m_l = l$ | (f) $\langle Y_{l,0} T Y_{l,0} \rangle > \langle Y_{l',0} T Y_{l',0} \rangle$ if $l < l'$ |
| (c) For each value of l there are $2l$ possible values of m_l | (g) (b) and (f) |
| (d) $L_+ Y_{l,l} = 0$ | (h) All of the above |

NAME: _____

4. What is the eigenvalue of L_z for Ψ if the eigenvalue of L^2 for Ψ is $25\hbar^2$ and the eigenvalue of $(L_x^2 + L_y^2)$ for Ψ is $16\hbar^2$?
- (a) The Heisenberg uncertainty principle dictates that Ψ cannot be an eigenfunction for L_z (e) $\pm 3\hbar$
- (b) $16\hbar^2$ (f) 0
- (c) $4i\hbar^2$ (g) π
- (d) $\pm 4\hbar$ (h) None of the above
5. For a diatomic rigid rotator having reduced mass 3 and bond length 2 a.u., which of the following statements is/are false?
- (a) The ground-state energy is zero (e) The moment of inertia is 12 a.u.
- (b) The energy separation between the first and second excited states is $(1/6)$ a.u. (f) (a) and (c)
- (c) The rotational constant B is $(1/24)$ a.u. (g) (b) and (e)
- (d) Transition from the ground state to the state $J = 3$ is forbidden (h) None of the above
6. For a spin-free hydrogenic wave function, which of the below relationships between quantum numbers is/are always true?
- (a) $n = l > m_l$ (e) $n > l \geq m_l$
- (b) $n = l + m_l$ (f) (b) and (c)
- (c) $n > l + m_l$ (g) (b) and (e)
- (d) $n > l > m_l$ (h) None of the above

NAME: _____

7. Which of the below statements about electron spin is/are true?
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|--|---|
| (a) The spin quantum number comes from including gravity in the electronic Schrödinger equation | (e) Spin-orbit coupling is proportional to the 8th power of the atomic number |
| (b) Spin couples with orbital angular momentum according to $\mathbf{J} = \mathbf{L} + \mathbf{S}$ | (f) (a) and (c) |
| (c) For a single electron, the only eigenvalues of S_z are $\pm(1/2)\hbar$ | (g) (a), (c), and (d) |
| (d) Stern and Gerlach discovered electron spin by studying the magnetic moments of Au atoms | (h) All of the above |
8. An electron of spin α is in a 5f orbital. Which of the below sets of quantum numbers (n, l, m_l, m_s) might describe such an electron?
- | | |
|--------------------|---------------------------|
| (a) (5, 4, 4, 4) | (e) (4, 4, 5, 1/2) |
| (b) (4, 3, 2, 1/2) | (f) (c) and (e) |
| (c) (5, 3, 0, 1/2) | (g) (b), (c), (d) and (e) |
| (d) (4, 5, 0, 7/2) | (h) None of the above |
9. What is the ground-state ionization potential for a one-electron atom having atomic number Z ?
- | | |
|--|---|
| (a) Z^2 a.u. | (e) 1 a.u. |
| (b) The negative of the energy of the electron in the 1s orbital | (f) The energy required to infinitely separate the nucleus and electron |
| (c) $(1/2)Z^2$ a.u. | (g) (b), and (d) |
| (d) $2Z^2$ a.u. | (h) (b), (c), and (f) |

NAME: _____

10. Which of the following wave functions has the greatest degeneracy?
- | | |
|--|--|
| (a) Particle in a box, level $n = 8$ | (e) Spin-free hydrogenic wave function, $n = 4$ |
| (b) Rigid rotator, $l = 4$ | (f) Relativistic free electron at rest |
| (c) Quantum mechanical harmonic oscillator, level $n = 25$ | (g) (b) and (e) have the same degeneracy which is greater than all of the others |
| (d) Spin-free hydrogenic wave function, $n = 6, l = 1$ | (h) (a) through (f) are all singly degenerate |

Short answer. Show that by proper choice of a , the function e^{-ar^2} is an eigenfunction of the operator

$$\left[\frac{d^2}{dr^2} - qr^2 \right]$$

where q is a constant. What is the name of the general class of functions represented by e^{-ar^2} ? How many nodes does this function have over r ?

NAME: _____

NAME: _____

Some Potentially Useful Mathematical Formulae

Trigonometric Relations

$$\sin \alpha \sin \beta = \frac{1}{2} [\cos(\alpha - \beta) - \cos(\alpha + \beta)]$$

$$\cos \alpha \cos \beta = \frac{1}{2} [\cos(\alpha - \beta) + \cos(\alpha + \beta)]$$

$$\sin \alpha \cos \beta = \frac{1}{2} [\sin(\alpha - \beta) + \sin(\alpha + \beta)]$$

$$\sin(\alpha \pm \beta) = \sin \alpha \cos \beta \pm \sin \beta \cos \alpha$$

$$\cos(\alpha \pm \beta) = \cos \alpha \cos \beta \mp \sin \alpha \sin \beta$$

$$\frac{d}{dx} \sin x = \cos x$$

$$\frac{d}{dx} \cos x = -\sin x$$

Integrals

$$\int_0^L \sin\left(\frac{m\pi x}{L}\right) \sin\left(\frac{n\pi x}{L}\right) dx = \frac{L}{2} \delta_{mn}$$

$$\int x \cos(ax) dx = \frac{x}{a} \sin ax + \frac{\cos ax}{a^2}$$

$$\int x^2 \cos(ax) dx = \frac{2x \cos ax}{a^2} + \frac{a^2 x^2 - 2}{a^3} \sin ax$$

$$\int_0^\infty r^n e^{-2r} dr = \frac{n!}{2^{n+1}}$$

Some Operators

$x \equiv$ multiply by x

$\mathbf{r} \equiv$ multiply by \mathbf{r}

$$p_x \equiv -i\hbar \frac{d}{dx}$$

$$H = T + V$$

$\boldsymbol{\mu} \equiv e\mathbf{r}$

$$\mathbf{L} \equiv \begin{vmatrix} \mathbf{i} & \mathbf{j} & \mathbf{k} \\ x & y & z \\ -i\hbar \frac{\partial}{\partial x} & -i\hbar \frac{\partial}{\partial y} & -i\hbar \frac{\partial}{\partial z} \end{vmatrix}$$

$$L_+ = L_x + iL_y \quad \text{and} \quad L_- = L_x - iL_y$$

Complex Relations

$$\sqrt{-1} = i = -\frac{1}{i}$$

$$e^{i\theta} = \cos \theta + i \sin \theta$$

Spherical Polar Volume Element

$$r^2 dr \sin \theta d\theta d\phi$$

NAME: _____